

Magnetic Interaction between Absorbed O₂ Molecules in [Cu₂(bza)₄(pyz)]_n

Minoru Soda and Takatsugu Masuda

Neutron Science Laboratory, Institute for Solid State Physics, University of Tokyo,

Tokai, Ibaraki 319-1106, Japan

Recently, [Cu₂(bza)₄(pyz)]_n (bza=benzoate, pyz=pyrazine), which can generate stable O₂-inclusion crystals at low temperature using forcible adsorption conditions, has been extensively investigated. [1,2] In this system, interesting magnetization was found, and the trimer induced by magnetic interactions between absorbed O₂ molecules in the nano-porous cavity was expected. In the present work, we carried out neutron scattering measurements on [Cu₂(bza)₄(pyz)]_n to clarify the magnetic interaction between absorbed O₂ molecules.

Neutron measurements were carried out using the cold neutron multi-chopper spectrometer Let installed at ISIS. After measurements on the [Cu₂(bza)₄(pyz)]_n polycrystalline sample sealed with high pressure oxygen gas (60 atm at room temperature), the measurements on the [Cu₂(bza)₄(pyz)]_n having no O₂ were also performed.

$S(Q, E)$ measured at 1.8 K for [Cu₂(bza)₄(pyz)]_n polycrystalline sample sealed with 60 atm oxygen gas indicates the dispersion-less excitations at $E \sim 0.4$ meV and 4 meV. Since these excitations were not observed for the [Cu₂(bza)₄(pyz)]_n having no O₂, these are originated from the magnetic moments of O₂. The constant- E scans show that both excitations have the peaks at $Q \sim 1.1$ Å.

We study the observed magnetic excitations using the trimer model shown in Fig. 1

$$H = J_1 S_1 \cdot S_2 + J_2 S_2 \cdot S_3 + D \{ (S_1^z)^2 + (S_2^z)^2 + (S_3^z)^2 \}.$$

As the result of the fitting, we obtained the exchange interactions $J_1 = 3.98$ meV, $J_2 = 0.79$ meV, and the uniaxial anisotropy $D = 0.44$ meV. The magnetic excitation can be reasonably explained by the trimer model. Although there is the large difference between the values of J_1 and J_2 , the similar difference has been reported in the theory about the Rh-system having the randomness of S_2 site.[2]

On the other hand, the temperature dependence of observed excitations is no simple. At this moment, we try to make the analyses of the T -dependence in order to clarify the magnetic correlation of trimer.

[1] S. Takamizawa, E. Nakata, and T. Akutsuka, *Angew. Chem., Int. Ed.* **45** (2006) 2216.

[2] S. Takamizawa et al., *J. AM. CHEM. SOC.* **130** (2008) 17883.

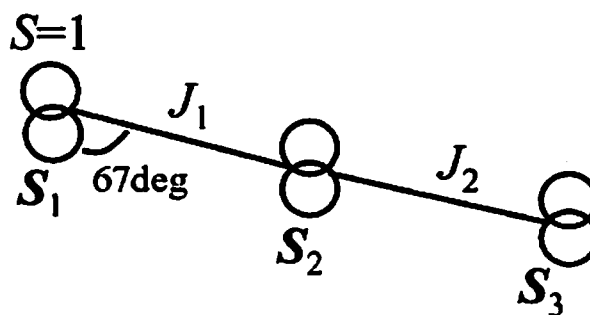


Fig. 1 Trimer model of oxygen molecules.